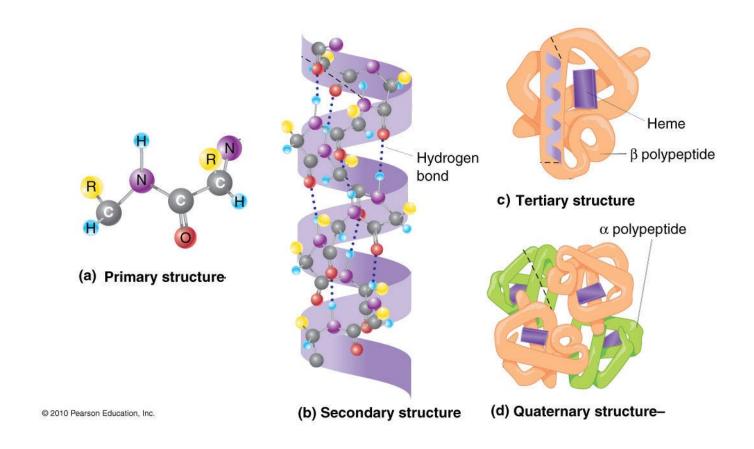
Learning Outcome 2 Understand the properties of nutrients



- 2.1 Explain how nutrients are structured
- 2.2 Classify the nutrients in food
- 2.3 Assess the impact of food production methods on nutritional value

Name	Form
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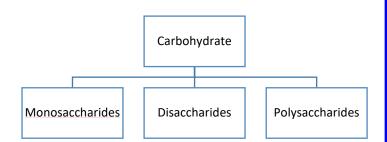
Proteins

Proteins are long chains of w	hich are joined by bonds. This forms
the structure of protein. These long chair	ns all contain the elements carbon, hydrogen,
; some proteins co	ntain the elements and
The secondary structure of	protein is when the chain to form a
helix or pleated sheet, this is held in place by	bonds. The
structure is where the chain folds on itself	f to form a structure, this is held in place by
a of chemical bonds. The	structure is where or
more proteins join together.	
Oxygen Amino acids variety hydrogen phosphorous sulphur tertiary 3D quarternary	primary weak covalent two nitrogen folds
Amino Acids	
The structure of an amino acid looks like this:	
THE STRUCTURE OF ALL ATTIMOS ACID TOOKS TIKE CHIS.	
There is a carboxyl group (COOH) at one end and an ar	mino group (NH2) at the other end of the molecule,
the R group	
Amino acids bond like this:	
This is known as a condensation reaction because	
The link between the 2 amino acids is known as a	
When 2 amino acids join together they are called a	when many amino acids
join together they are called	
Examples of proteins are	which is found in red blood cells, elastin, which is
found in ligaments, collagen which is found in muscle,	which is found in eggs and
which is a hormone that regul	ates sugar in the blood.

Carbohydrates

These are made up of the elements , and and

Photosynthesis



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Polysaccharides include

Glucose is a simple sugar found in

It is a six sided molecule (hexose) with 6 carbon atoms in a ring

Glucose molecules join like this:

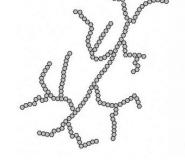
Dissacharides are double sugars – made when 2 monosaccharides join together. They include:

Glucose + Glucose = Glucose + Fructose = Glucose + Galactose =

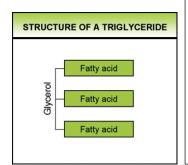
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Lipids

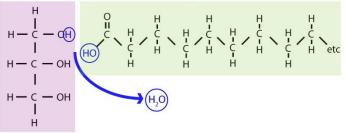


Lipids are made up of one unit of and 3 and 3

Fatty acids can be saturated in which case they look like this: -C-C-C-C-C-H

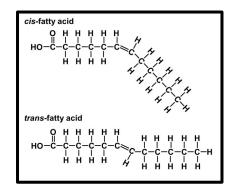
Or unsaturated in which case they look like this: : -C-C-C-C-C-H

Glycerol Fatty acid



When a fatty acid joins to a unit of glycerol a unit of water is produced (H₂0) so it is called areaction.

Each of the 3 fatty acids joining the glycerol is different which means that the fat is 'plastic' ie it melts over a range of temperatures, this gives the fat the property of being



Cis fatty acids
Trans fatty acids

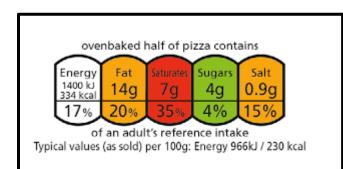
The term 'saturated means
Examples of saturated fats are
The term monounsaturated means
Examples of monounsaturated fats are
The term polyunsaturated means
Examples of polyunsaturated fats are

Explain the difference between saturated and unsaturated fats giving examples

Classify the nutrients in food

In this section we need to consider the different ways we can classify the nutrients in food.

Explain the following terms:
Biological value
Glycemic Index
Nutrient Density
Complementary action of nutrients (and give examples)



How does a food label help a consumer to classify/understand the nutrients in this product?



How does this information help consumers to make healthy choices?

How could consumers use recipes to inform them about the nutrients in their food?

Impact of production methods on nutritional value

Discuss how the following methods affect the nutritive value of foods, consider both micro and macro nutrients.

Boiling		Roasting	
Steaming		Deep fat frying	
How do the following storage	e methods affect the nu	utritional value of fo	ond?
		difficital value of to	
Vacuum packaging	Cold Store		Aseptic food processing and packaging (AFP)
How do the following preserv	ation methods affect t	he nutritive value o	f food?
Freezing	Jamming		UHT
What is meant by the term fo	ortification?		
What foods are fortified by la	aw?		
What foods are fortified volu	ntarily ?		

Glossary of terms

Fortification

Glycemic Index
Biological value
Nutrient density
Lipids
Proteins
HBV
LBV
Carbohydrates
NSP
Soluble fibre
Insoluble fibre
Traffic light labels
Condensation reaction
Monosaccharide
Disaccharide
Polysaccharide
Amino Acid
Dipeptide
Polypeptide

Exam Style Questions

- 1. What are the arguments for and against the fortification of foods, give examples to support your answers? (6 marks)
- 2. Compare the chemical structure and characteristics of simple and complex carbohydrates (6 marks)
- 3. Describe the effects of cooking on the water soluble vitamins giving examples in your work (6 marks)
- 4. Explain the meaning of the following terms:
 - a. Glycemic index (2 marks)
 - b. Biological value (2 marks)
- 5. Compare roasting and deep fat frying on the nutrient content of foods (4 marks)
- 6. Explain the difference between saturated and polyunsaturated fats (6marks)
- 7. Explain how simple sugars join to make polysaccharides (6 marks)
- 8. Assess how food production methods impact on the nutritional value of food products, giving examples (8 marks)
- 9. Which foods are fortified
 - a. By law
 - b. Voluntarily
- 10. Why would a food manufacturer choose to fortify foods?
- 11. Explain the complimentary action of nutrients
 - a. LBV proteins
 - b. Calcium and vitamin D
 - c. Iron and vitamin C
- 12. Explain how jamming can affect the nutritional value of a food (4 marks)
- 13. Describe, using a diagram, how amino acids join together to make proteins
- 14. Name the most common proteins in Eggs, meat, milk
- 15. Explain the impact of freezing on the nutritional value of foods
- 16. What does the term nutrient density mean and how can it help consumers to choose a healthy diet?
- 17. What is the R group in an amino acid?
- 18. Describe how fatty acids and glycerol bond to make a triglyceride
- 19. Explain the complementation of protein, giving a variety of examples
- 20. How do traffic light labels help consumers to make healthy choices?