	ORE KNOWLEDGE	What I will know and understand by the end of Year 7.		
Du	ring this year in science we will be developing our scientific knowledge and conceptual understanding of-	This links to:	Key Voc	abulary:
1	<ul> <li>Organisation - biological molecules that make up the diet, energy in food, the skeleton, the digestive system, reproduction, gametes, fertilisation, pregnancy and birth.</li> <li>Energy - energy stores and transfers, types of energy resources.</li> <li>Particle model - the particle, solids, liquids and gases, changes in state, physical and chemical changes.</li> </ul>	<ul> <li>Y4 &amp; Y6 digestive system and nutrients.</li> <li>Introduction to energy.</li> <li>Y4 &amp; Y5 properties of different materials.</li> </ul>	Carbohydrate Protein Digestion Enzyme Gamete	Fertilisation Renewable Non- Renewable Physical Chemical
2	<ul> <li>Assessment Point 1 - All content from half term 1.</li> <li>Cells and Transport - levels of organisation, animal cells and organelles, diffusion.</li> <li>Variation - Watson, Crick and Franklin, DNA, genes, chromosomes and heredity.</li> <li>Chemistry of the Atmosphere- structure of the earth, the atmosphere, rocks and the rock cycle.</li> <li>Bioenergetics - aerobic respiration.</li> </ul>	Introduction to animal cells. Y1,Y4 & Y6 how organisms are grouped together. Y3 types of rock. Cells - the role of the mitochondria.	Organelle Mitochondria Ribosome Respiration Nucleus	Chromosome Atmosphere Igneous Metamorphic Sedimentary
3	<ul> <li>Cells and Transport - plant cells and organelles.</li> <li>Bioenergetics - photosynthesis and plant adaptations.</li> <li>Atomic Structure - introduction to atoms, elements, compounds and mixtures.</li> <li>Chemical Changes - the rearrangement of atoms in a chemical reaction, chemical formulae and equations.</li> <li>Electricity and magnetism - static electricity, circuit symbols, permanent magnets.</li> <li>Waves - longitudinal and transverse waves, the human ear, amplitude, frequency and ultrasound.</li> </ul>	HT2 Animal cells Plant cells - the role of the chloroplast Introduction to atomic structure. Y5 formation of new materials being irreversible. Y4&Y6 electric circuits. Y3,Y4,Y6 light and sound waves.	Chloroplast Vacuole Photosynthesis Glucose Atom	Element Compound Series Parallel Current
4	<ul> <li>Assessment Point 2 - All content from half terms 1-3.</li> <li>Atomic Structure-States of Matter and Atomic Model. Changes of state, chemical and physical properties of elements, metals and non metals, the periodic table, chemical symbols and formulae, atomic model and subatomic particles.</li> <li>Space - seasons, the solar system, stars and galaxies.</li> </ul>	HT1 particle model,HT3 atomic structure. Y1,Y5 the sun and planets.	Galaxy Universe Gravity Orbit	Properties Proton Neutron Electron Symbol
5	<ul> <li>Atomic Structure - Separating Mixtures, diffusion, filtration, evaporation, distillation and chromatography.</li> <li>Forces - contact and non-contact forces, balanced and unbalanced forces, air resistance, friction and gravity.</li> </ul>	HT3 & HT4 atomic structure. Y3 & Y5 contact and non contact forces.	Evaporation Distillation Filtration Chromatography Diffusion	Resultant Air resistance Friction Balanced
6	<ul> <li>Assessment Point 3 - All content from half terms 1-5.</li> <li>Ecology - food chains, food webs, accumulation of toxic substances and sampling techniques.</li> <li>Rates of Reaction - collision theory, the effect of temperature and catalysts.</li> <li>Energy Changes - exothermic and endothermic reactions.</li> </ul>	Y1,Y2,Y4 habitats and food chains. HT3 chemical changes. HT3 chemical changes & HT6 rates of reaction.	Organism Accumulation Quadrat Sampling Ecosystem	Collision Catalyst Temperature Exothermic Endothermic

We will learn to use these command words; state, describe, calculate, define, choose, draw, give, identify, label, name, write We will learn to develop these practical skills; control variables, risk assessments, obtaining and analysis of data.

Target Grade: AP1:	AP2:	AP3:	
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## CORE KNOWLEDGE

## What I will know and understand by the end of Year 8.



During this year in science we will be developing our scientific knowledge and conceptual understanding of-		This links to:	Key Vocabulary:			
1	<ul> <li>Cells and Transport - using microscopes, diffusion and osmosis, specialised cells.</li> <li>Quantitative Chemistry - relative atomic/formula mass, conservation of mass.</li> <li>Atomic Structure - the periodic table, group 1 and group 7.</li> <li>Energy - energy changes and transfers, conduction, convection, radiation, power, fuel costs.</li> </ul>	<ul> <li>Y7 HT2 &amp; HT3 Cells</li> <li>Y7 HT 4 atomic model HT3 chemical changes.</li> <li>Y7 HT4 atomic model.</li> <li>Y7 HT1 energy</li> </ul>	Light microscope Osmosis Specialised cells Differentiation. Conservation Properties	Groups Periods Energy stores Conduction Convection Radiation		
2	<ul> <li>Assessment Point 1 - All content from half term 1 + Year 7 content.</li> <li>Infectious Disease - unicellular organisms such as bacteria a viruses.</li> <li>Bonding - ions and ionic bonding.</li> <li>Particle Model - diffusion and Brownian motion.</li> <li>Chemical Changes - pH, reactions of metals, displacement, thermal decomposition, oxidation, combustion.</li> </ul>	Introduction to infectious disease. Introduction to bonding - links to atomic structure. Y7 HT1 particle model Y7 HT3 chemical changes	Unicellular Bacteria Virus Ions Diffusion	pH Neutralisation Combustion Displacement Extraction		
3	<ul> <li>Organisation - gut bacteria, nutrient deficiencies, muscles, circulatory system, lungs, drugs.</li> <li>Electricity and Magnetism - resistance, series and parallel circuits and electromagnets.</li> <li>Energy Changes - bond energies and activation energy.</li> </ul>	<ul> <li>Y7 HT1 organ systems</li> <li>Y7 HT3 circuit symbols</li> <li>Y7 HT6 exothermic and endothermic reactions.</li> </ul>	Deficiency Obesity Antagonistic Breathing Recreational	Resistance Ohm's Law Electromagnets Bond energy Activation energy		
4	<ul> <li>Assessment Point 2 - All content from half terms 1-3 - Y7 content.</li> <li>Bioenergetics - anaerobic respiration.</li> <li>Space - gravity on different planets, the light year.</li> <li>Bonding - covalent bonds and simple covalent molecules.</li> <li>Organic Chemistry - crude oil and alkanes.</li> </ul>	<ul> <li>Y7 HT2 aerobic respiration.</li> <li>Y7 HT4 the solar system</li> <li>Y8 HT2 ionic bonding</li> <li>Introduction to organic chemistry</li> </ul>	Anaerobic Lactic Fermentation Light year Weight	Covalent Crude oil Hydrocarbon Alkane		
5	<ul> <li>Organisation - plant reproduction, pollination.</li> <li>Waves - human eye, absorption and transmission, refraction, colour.</li> <li>Chemistry of the atmosphere - carbon cycle, composition of the atmosphere, global warming.</li> </ul>	<ul> <li>Y3 life cycle of flowering plants</li> <li>Y7 HT3 transverse and longitudinal waves.</li> <li>Y7 HT2 the Earth's Atmosphere</li> </ul>	Pollination Fertilisation Dispersal Stigma Stamen	Carbon cycle Composition Climate Weather		
6	<ul> <li>Assessment Point 3 - All content from half terms 1-5 + Y7 content.</li> <li>Variation and Ecology - variation, natural selection, extinction, biodiversity.</li> <li>Forces - speed, distance- time graphs, Hooke's Law, pressure in liquids.</li> <li>Using resources - recycling, polymers, composites, ceramics</li> </ul>	<ul> <li>Y7 HT2 genes and DNA Y7 HT6 feeding relationships</li> <li>Y7 HT5 contact and non contact forces.</li> <li>Introduction to using resources.</li> </ul>	Continuous Discontinuous Species Natural selection Evolution	Extinction Moments Pressure Polymer Ceramics		
We will learn to use these command words; state, describe, calculate, define, choose, draw, give, identify, label, name, write, compare, explain, determine, justify, plan, predict, evaluate, suggest.						

We will learn to develop these practical skills; control variables, risk assessments, obtaining and analysis of data, planning.

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CORE KNOWLEDGE What I will know and understand by the end of Year 9.								
During this year in science we will be developing our scientific knowledge and conceptual understanding of-			This l	inks to:	Key Vocabulary:			
1	<ul> <li>Separatio periodic to Conservat</li> <li>Assessment F</li> <li>Space</li> </ul>	periodic table and its history. • Conservation of mass, balancing equations, RFM and RAM. Assessment Point 1 - Atomic structure + Y7 & Y8 content. Space Y7 HT5 the solar			HT1 quantitative chem Y7 HT5 the solar syst		Proton Neutron Electron Isotope RAM	Conservation RFM Supernova Satellite Lifecycle
2	<ul> <li>Cells and Transport         <ul> <li>Cell structure and function, organelles, magnification, transport of substances, cell division, stem cells RP - Magnification, RP - Osmosis.</li> </ul> </li> <li>Particles Model-         <ul> <li>Density, changes of state, internal energy and particle motion. RP- Density.</li> </ul> </li> </ul>			<ul> <li>Y8 HT1 microscopes, specialised cells.</li> <li>Y7 HT1 the particle, s</li> </ul>	<ul> <li>2&amp;3 animal and plant cells.</li> <li>1 microscopes, diffusion, osmosis &amp; ised cells.</li> <li>1 the particle, states of matter, s of state. Y8 HT2 Brownian motion.</li> </ul>		Chromosome Differentiation Density Motion Irregular	
3	<ul> <li>Assessment Point 2 All Y9 HT1-2 content &amp; synoptic section.</li> <li>Bonding</li> <li>Metallic bonding, alloys, ionic bonding, giant ionic structures, covalent bonding, simple covalent molecules, giant covalent structures and polymers.</li> </ul>			<b>Y8 HT2</b> ions and ionic bonding <b>Y8 HT4</b> covalent bonding		Metallic Covalent Ionic Properties Structure	Ion Delocalised Graphite Diamond Polymer	
4	<ul> <li>Energy Changes</li> <li>Endothermic and exothermic reactions and bond energies. RP - Energy changes.</li> <li>Atomic Structure (physics)</li> <li>Structure of the atom and its development, nuclear radiation and decay, half life.</li> </ul>				<ul> <li>Y7 HT6 exothermic and endothermic reactions. Y8 HT3 bond energies and activation energy.</li> <li>Y7 HT4, Y8 HT1, Y9 HT1 chemistry atomic structure.</li> </ul>		Endothermic Exothermic Enthalpy Bond energy	Alpha Beta Gamma Half life Radioactive
5	<ul> <li>Assessment Point 2 All Y9 HT1-4 content &amp; synoptic section.</li> <li>Chemical Changes <ul> <li>Redox reactions, metals and acids, extraction of metals, neutralisation and electrolysis. RP- Making salts, RP-Electrolysis.</li> </ul> </li> </ul>				<b>Y7 HT3</b> rearrangement of atoms, <b>Y8 HT</b> reaction of metals, oxidation and displacement.		Oxidation Reduction Extraction Neutralisation	Electrolysis Ion Salt Solution
6	<ul> <li>Infectious Disease</li> <li>Viruses, bacteria, fungal diseases, malaria, white blood cells, vaccinations, antibiotics, medicines and development of new drugs.</li> </ul>			<b>Y9 HT2</b> cells <b>Y8 HT2</b> bacteria and viruses.		Virus Bacteria Protist Vaccination Antibodies	Antibiotics Lymphocytes Phagocytes Phagocytosis	
Tar	Target Grade:   AP1:			AP2:		AP3:		