



SCIENCE |

COMBINED SCIENCE: TRILOGY

We are following the **AQA Combined Science: Trilogy** specification. Assessment consists of 6 papers altogether, two biology, two chemistry and two physics, each will assess different topics.

Duration: all the papers are 1 hour 15 minutes.

Tiers: Foundation and Higher.

Weighting: Each paper consists of 70 marks and is worth 16.7% of the grade.

Question types: multiple choice, structured, closed, short answer and open response. 15% of GCSE marks in exams come from questions relating to practicals.

Combined Science will have a 17 point grading scale, from 9–9, 9–8 through to 2–1, 1–1.

SEPARATE SCIENCES

We are following the **AQA specifications**. Each science is assessed separately, leading to the award of three separate GCSEs. For **each** of the sciences assessment consists of 2 papers: each paper will assess knowledge and understanding from different topics.

Duration: both papers are 1 hour 45 minutes. **Tier:** Foundation and Higher.

Weighting: the papers are equally weighted. Each is worth 50% of the grade and has 100 marks available.

Question types: multiple choice, structured, closed short answer and open response. 15% of GCSE marks in exams come from questions relating to practicals.

The A* to G grades will be replaced by 9 to 1 for Biology, Chemistry and Physics

EXAM DATES

Combined Science and Triple Science

Biology	Paper 1/1	B1-4	Tuesday 12 th May 2020
Chemistry	Paper 2/1	C1-5	Thursday 14 th May 2020
Physics	Paper 3/1	P1-4	Wednesday 20 th May 2020
Biology	Paper 4/2	B5-7	Monday 1 st June 2020
Chemistry	Paper 5/2	C6-10	Wednesday 10 th June 2020
Physics	Paper 6/2	P5-7 or 8	Friday 12 th June 2020

REVISION GUIDES

Purchase a science specific revision guide
(from school - if you haven't already done so)

Comes with
free online
access

Functions of the Blood

Blood is very useful stuff. It's a big transport system for moving things around the body. The **blood cells** do good work too. The **red blood cells** are responsible for transporting **oxygen** about, and they carry 100 times more than could be moved just dissolved in the plasma. And as for the white blood cells...

Plasma is the Liquid Bit of Blood

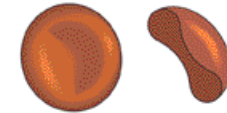
It's basically blood minus the blood cells (see below). Plasma is a pale yellow liquid which carries just about everything that needs transporting around your body:

- 1) **Red and white blood cells** (see below) and **platelets (used in clotting)**.
- 2) **Water**.
- 3) Digested food products like **glucose** and **amino acids** from the gut to all the body cells.
- 4) **Carbon dioxide** from the body cells to the lungs.
- 5) **Urea** from the liver to the kidneys (where it's removed in the urine).
- 6) **Hormones** — these act like chemical messengers.
- 7) **Antibodies** and **antitoxins** produced by the white blood cells (see below).

Red Blood Cells Have the Job of Carrying Oxygen

They transport **oxygen** from the **lungs** to **all** the cells in the body. The **structure** of a red blood cell is adapted to its **function**:

- 1) Red blood cells are **small** and have a **biconcave shape** (which is a posh way of saying they look a little bit like doughnuts, see diagram below) to give a **large surface area** for **absorbing** and **releasing oxygen**.
- 2) They contain **haemoglobin**, which is what gives blood its **colour** — it contains a lot of **iron**. In the lungs, haemoglobin **reacts with oxygen** to become **oxyhaemoglobin**. In body tissues the reverse reaction happens to **release oxygen to the cells**.
- 3) Red blood cells don't have a **nucleus** — this frees up **space** for more haemoglobin, so they can carry more oxygen.



White Blood Cells are Used to Fight Disease

...against disease.

to fight microbes.

to neutralise the toxins produced by microbes.

...which helps them to **engulf** any micro-organisms they come across. Basically the white blood cell wraps around the micro-organism and, and then it **digests it** using enzymes.

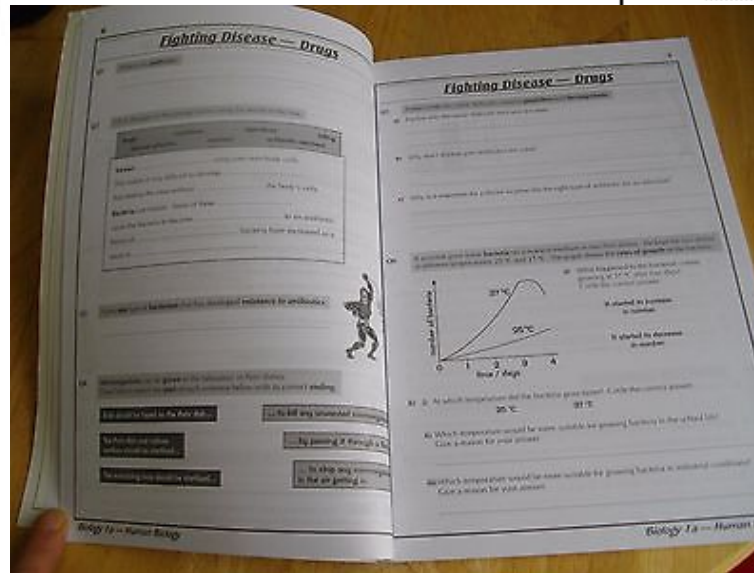
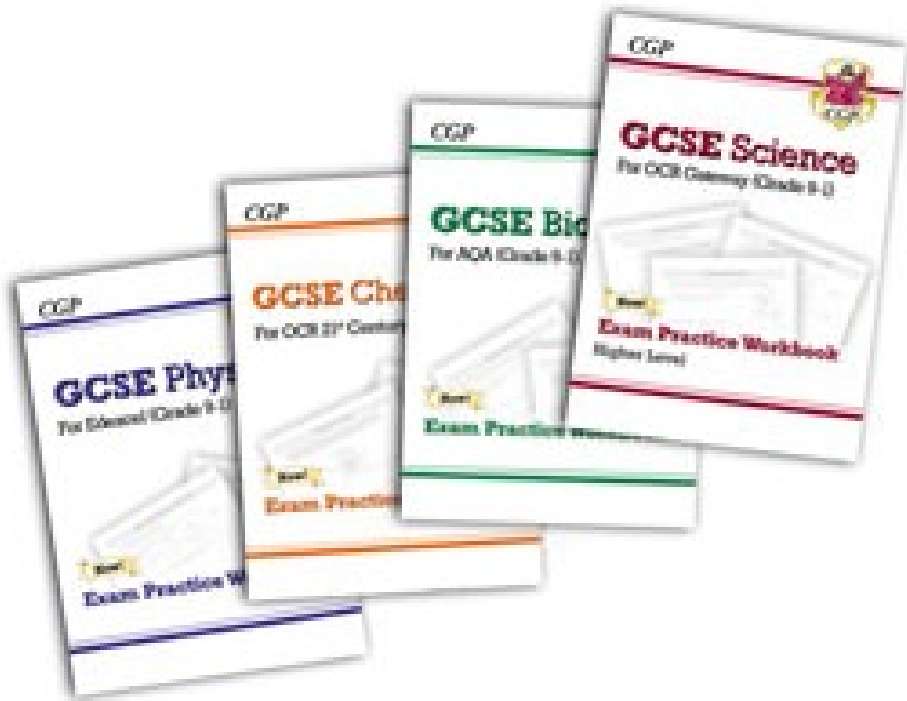


Sweat and tears — kind of... ...without the sweat... or the tears... just the blood then... yep... anyway... contains about **six and a half pints** of blood altogether, and every single drop of blood there are usually about 500 times more red blood cells than white.

and Growing

REVISION WORKBOOKS

Some students may benefit from using a dedicated science specific workbook (available with answer booklet from the school shop)



Static Electricity

Q1 Circle the pairs of charges that would attract each other and underline those that would repel.

positive and positive positive and negative negative and positive negative and negative

Q2 Fill in the gaps in these sentences with the words below.

electrons	positive	static	insulating	negative
..... electricity can build up when two materials are rubbed together. The move from one material onto the other. This leaves a charge on one of the materials and a charge on the other.				

Q3 The sentences below are wrong. Write out a **correct** version for each.

a) An insulating rod becomes negatively charged when rubbed with a duster because it loses electrons.

Polythene is an insulating material.

..... and polythene rod will repel small pieces of paper if they are placed near it.

..... er two charged objects are together, the less strongly they attract or repel.

..... ively charged object is connected to earth by a metal strap, flow through the strap from the object to the ground.

..... of static can cause sparks if the distance between the object and the earth is big enough.

Google drive – revision materials

The screenshot displays a Google Drive web interface within a browser window. The browser's address bar shows the URL <https://drive.google.com/drive/folders/17k8dWGnkhRbjs82Fo7qwyV0vSN7Unorp>. The page header includes the BGLC Drive logo, a search bar, and a breadcrumb trail: **GCSE Science Revision > Combined Science (Trilogy) > Biology Paper 1 > B1 Cell Biology**. The left sidebar contains navigation options: My Drive, Team Drives, Shared with me, Recent, Google Photos, Starred, and Bin, with a status of 5 GB used. The main content area, titled 'Files', displays a grid of document thumbnails. These include two PDFs titled 'B1 Cell Biology R...', a Word document 'B1 Complete You...', a document 'B1 low demand p...', a document 'B1 standard dem...', a Word document 'B1,B2 & B3 Past ...', and a presentation 'Biology-Revision-...'. The bottom of the image shows a Windows taskbar with the search bar 'Type here to search', various application icons, and a system clock indicating 17:02 on 06/01/2018.

(b) When sodium chloride solution is electrolysed the products are hydrogen and chlorine.

(i) What is made from chlorine?

Tick (✓) **one** box.

Bleach

☐

Fertiliser

☐

Soap

☐

(ii) Sodium chloride solution contains two types of ion: sodium ions (Na^+) and chloride ions (Cl^-).

Why is hydrogen produced at the negative electrode?

Tick (✓) **one** box.

3. I can calculate total magnification. ☐

3. I can calculate total magnification. ☐

Describe the structure of a plum pudding model.

Alpha particles are made of two protons and two neutrons.

What the formula means


- HCl
 - Each capital letter shows a new element
 - If there are no numbers this means there is one of each
 - 1 H and 1 Cl
- CaCO_3
 - A little number only applies to the number it follows
 - 1 Ca, 1 C but 3 O's
- 2KI
 - A big number means that it applies to everything after it
 - So this has 2 K's and 2 I's
- $\text{Mg}(\text{OH})_2$
 - Numbers in the brackets times by the number outside the brackets
 - So this has 1 Mg, 2 O and 2 H

Neutron relative mass:

Electron relative charge






← → ↻ Secure | https://app.:

SENECA BETA

 **Biology: AQA GCSE Higher**

2 Organisation

- 2.1 Principles of Organism...
- 2.2 Enzymes
- 2.3 Circulatory System
 - 2.3.1 Blood Vessels
 - 2.3.2 Blood Vessels 2
 - 2.3.3 The Heart**
 - 2.3.4 Circulatory System & Gas Exchange
 - 2.3.5 Blood
 - 2.3.6 Blood Cells
- 2. Share Free Teacher CPD Course


2.     


YOUTUBE RECOMMENDED CHANNELS


HELPFUL CHANNELS FOR REVISION TIPS OR SCIENCE TOPICS

Below are a list of recommended channels on YouTube that have videos that would be helpful in your Science revision:

Youtubers recommended for Science topics and revision tips:

Revision with Eve  Revision With Eve
94,778 subscribers

Primrose Kitten  Science and Maths by Primrose Kitten
38,981 subscribers

Christopher Thornton  Christopher Thornton
32,745 subscribers

Youtubers recommended for Science topics:

My GCSE Science  myGCSEscience
17,022 subscribers

Free Science Lessons  Free Science Lessons
19,074 subscribers

Youtubers recommended for Revision skills:

Study with Jess  Study With Jess
39,828 subscribers

cular organ that pumps
ly.
ambers: the left and
t and right ventricles.
s and a pacemaker.
he body's left, but the
eart from a doctor's

GCSE GOOGLE REVISION CLASSROOM

Mr Dixey has invited all students to the revision classroom, please encourage your child to participate.

Questions will be posted on various topic weekly
(answers the following week)

REVISION IN SCHOOL

Most classes will finish formal teaching of content by mid march

Mock exams shortly after covering the most recent modules

Approx. 6 weeks for revision in class:

Re-teaching of difficult topics

Independent study

Focussed exam question prep

NO HEADPHONES SO DON'T ASK!



Introduction
The study of the relationship between the individual and the society is known as sociology.

Definition of
Sociology is the study of society and social behavior.

Classification
Sociology is classified into two main branches: **Primary** and **Secondary**.

Types of
Sociology is divided into two main types: **Primary** and **Secondary**.

Importance
Sociology is important for the study of society and social behavior. It helps us to understand the social structure and the social behavior of the individual.

Conclusion
Sociology is a social science that studies the relationship between the individual and the society. It is a branch of knowledge that helps us to understand the social structure and the social behavior of the individual.

AQA

SPECIMEN MATERIAL

GCSE
COMBINED SCIENCE: TRILOGY

H

Higher Tier Paper 1: Biology 1H

Specimen 2018

Time allowed: 1 hour 15 minutes

Materials

For this paper you must have:

- a ruler
- a calculator.

Instructions

- Answer all questions in the spaces provided.
- Do all rough work in this book. Cross through any work you do not want to be marked.

Information

- There are 70 marks available on this paper.
- The marks for questions are shown in brackets.
- You are expected to use a calculator where appropriate.
- You are reminded of the need for good English and clear presentation in your answers.
- When answering questions Q2.2, Q5.3 and Q6.6 you need to make sure that your answer:
 - is clear, logical, sensibly structured
 - fully meets the requirements of the question
 - shows that each separate point or step supports the overall answer.

Advice

- In all calculations, show clearly how you work out your answer.

Please write clearly, in block capitals.

Centre number Candidate number

Surname

Forename(s)

Candidate signature _____

EXAM TIPS

There are 21 formula
(23 for Separates) to
remember for the exam

@crumptonn

Trilogy - You must memorise all of these formulae

1	pressure due to a column of liquid = height of column \times density of liquid \times gravitational field strength (g)	$p = h \rho g$
2	(final velocity) ² - (initial velocity) ² = 2 \times acceleration \times distance	$v^2 - u^2 = 2 a s$
3	force = $\frac{\text{change in momentum}}{\text{time taken}}$	$F = \frac{m \Delta v}{\Delta t}$
4	elastic potential energy = 0.5 \times spring constant \times (extension) ²	$E_e = \frac{1}{2} k e^2$
5	change in thermal energy = mass \times specific heat capacity \times temperature change	$\Delta E = m c \Delta \theta$
6	period = $\frac{1}{\text{frequency}}$	$T = \frac{1}{f}$
7	magnification = $\frac{\text{image height}}{\text{object height}}$	

More difficult scientific
formula (Physics papers)
and a periodic table
(Chemistry papers) are
given in the exams

11452008

The Periodic Table of the Elements

Key
relative atomic mass
atomic symbol
atomic (proton) number

1	2	3	4	5	6	7	8
1 H 1.008							4 He 4.003
7 Li 6.941	9 Be 9.012						20 Ne 20.180
23 Na 22.990	24 Mg 24.305						36 Ar 35.963
39 K 39.098	40 Ca 40.078	45 Sc 44.956	48 Ti 47.88	51 V 50.942	52 Cr 51.996	55 Mn 54.938	58 Fe 55.845
85 Rb 85.468	88 Sr 87.62	89 Y 88.906	91 Zr 91.224	93 Nb 92.906	96 Mo 95.94	101 Ru 101.07	106 Pd 106.42
133 Cs 132.91	137 Ba 137.33	178 La* 175.05	174 Hf 178.49	181 Ta 180.95	186 W 183.84	192 Re 186.21	197 Os 190.23
223 Fr [223]	226 Ra [226]	227 Ac* [227]	228 Th [228]	232 Pa [232]	238 U [238]	244 Pu [244]	247 Am [247]

* The lanthanoids (atomic numbers 58-71) and the actinoids (atomic numbers 90-103) have been omitted.
The relative atomic masses of copper and chlorine have not been rounded to the nearest whole number.

$$= B I l$$

$$= m L$$

$$i = \frac{n_p}{n_s}$$

$$v = V_s / s$$

$$\text{constant}$$

Eqn	Word Equation	Symbol Equation
1	weight = mass \times gravitational field strength	$W = m \times g$
2	work done = force \times distance (along the line of action of the force)	$W = F \times s$
3	force applied to a spring = spring constant \times extension	$F = k \times e$
4		$s = v \times t$
5		$a = \frac{\Delta v}{t}$ or $a = \frac{(v - u)}{t}$
6		$F = m \times a$
7		$p = m \times v$
8		$E_k = \frac{1}{2} \times m \times v^2$
9	gravitational field strength \times height	$E_p = m \times g \times h$
10		$p = \frac{F}{A}$
11		$p = \frac{W}{t}$
12	efficiency = $\frac{\text{useful output energy transfer}}{\text{total input energy transfer}}$	
13	efficiency = $\frac{\text{useful power output}}{\text{total power input}}$	
14	wave speed = frequency \times wavelength	$v = f \times \lambda$
15	charge flow = current \times time	$Q = I \times t$
16	potential difference = current \times resistance	$V = I \times R$
17	power = potential difference \times current	$P = V \times I$
18	power = (current) ² \times resistance	$P = I^2 \times R$
19	energy transferred = power \times time	$E = P \times t$
20	energy transferred = charge flow \times potential difference	$E = Q \times V$
21	density = $\frac{\text{mass}}{\text{volume}}$	$\rho = \frac{m}{V}$

EXAM TIPS

Rough guide is 1 mark per minute!

Have a go, if in doubt put something it down (no answer = no mark)

9 Enzymes have many industrial uses.

(a) Draw straight lines to join each **enzyme** with the correct **use of the enzyme**.

Draw only **three** lines.

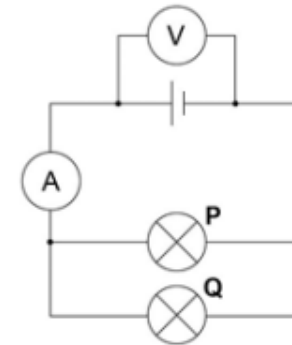
Enzyme	Use of the enzyme
sucrase	used in the production of milk for people with intolerance to dairy products
lactase	used on reagent strips to detect lactose
ligase	used to join strands of DNA together
	used to produce sweeter sugars for food

0 1

Figure 1 shows a circuit diagram containing two identical lamps arranged in parallel.

The reading on the ammeter is 186 mA.

Figure 1



0 1 . 1

Which statement about the current through the lamps is true?

[1 mark]

Tick **one** box.

The current through both lamp **P** and lamp **Q** is **0.093 A**

☐

The current through both lamp **P** and lamp **Q** is **0.186 A**

☐

The current through both lamp **P** and lamp **Q** is **0.93 A**

☐

The current through both lamp **P** and lamp **Q** is **1.86 A**

☐

LONGER ANSWER QUESTIONS

Don't be daunted by the 4 - 6 mark questions.

Read the stem of the question, it often has vital information.

Read the command words carefully – describe, explain, compare, evaluate

If data/graph is given, use it!

It is OK to bullet point your answer.

Read through what you have written!!!

Dare to have a go!

3 Look at the picture of a firefly.

The firefly is able to give out flashes of bright light to attract a mate.

Just after dark is the best time to see fireflies flashing light.

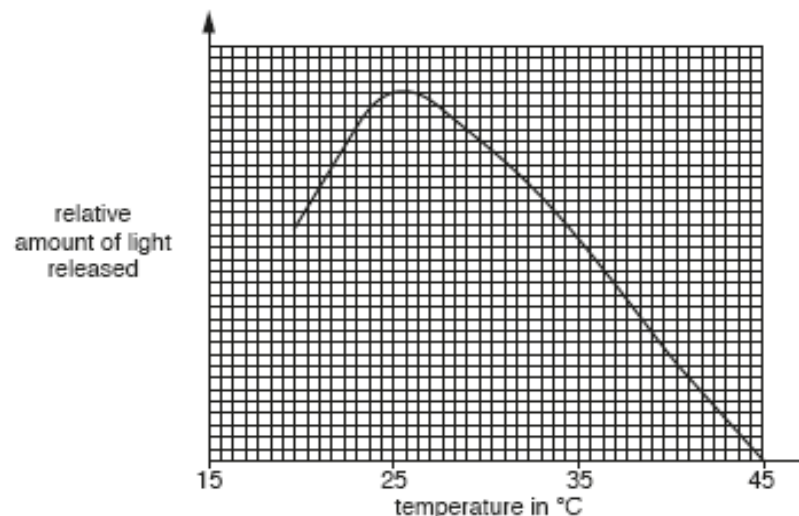


The reaction that releases the light involves the breakdown of a chemical.

An enzyme called luciferase is needed for this reaction.

Look at the graph.

It shows how temperature affects the reaction that releases light.



- (a) Use data from the graph to **explain** the effect of temperature on luciferase and explain why it is **only** luciferase enzyme that will catalyse this reaction.

If data is given, use it!

Mark Scheme

June 2015

Answer	Marks	Guidance
<p>[Level 3] Explains the effects of temperature on luciferase AND explains the specificity of enzymes. Quality of written communication does not impede communication of the science at this level. (5 – 6 marks)</p> <p>[Level 2] Explains the effects of temperature on luciferase OR explains the specificity of enzymes. Quality of written communication partly impedes communication of the science at this level. (3 – 4 marks)</p> <p>[Level 1] Describes the effects of temperature on luciferase AND describes the specificity of enzymes. Quality of written communication impedes communication of the science at this level. (1 – 2 marks)</p> <p>[Level 0] Insufficient or irrelevant science. Answer not worthy of credit. (0 marks)</p>	6	<p>This question is targeted at grades up to A.</p> <p><u>Levels 2 and 3</u> Indicative scientific points to explain specificity include:</p> <ul style="list-style-type: none"> • 'lock and key' mechanism named & explained • substrate shape matches (active site of) luciferase only • a different substrate shape does not match (active site of) luciferase <p>allow correctly labelled diagram showing 'lock' and key' ignore only luciferase enzyme catalyses this reaction (in question)</p> <p>Indicative scientific points to explain effects of temperature include:</p> <ul style="list-style-type: none"> • active site changes shape when denatured (so substrate won't fit) • denaturing may start to occur at around 28°C / occurs at any temperature above optimum • lower collision rates at temperatures around 15°C • higher collision rates at temperatures around 27°C <p><u>Level 1</u> Indicative scientific points to describe specificity include:</p> <ul style="list-style-type: none"> • enzymes only work with one substance • enzymes have an active site <p>Indicative scientific points to describe effects of temperature include:</p> <ul style="list-style-type: none"> • rate of reaction increases between 20°C and 26°C • rate of reaction decreases between 26°C and 45°C • optimum temperature quoted as 25°C or 26°C or 27°C • reaction stops at 45°C <p>allow rate increase or decrease anywhere within the range given above</p> <p>Use the L1, L2, L3 annotations in Scoris. Do not use ticks.</p>

0 5 . 3 In coronary heart disease (CHD) layers of fatty material build up inside the coronary arteries. This can cause a heart attack.

Statins and stents can be used to reduce the risk of a heart attack in people with CHD.

Evaluate the use of statins and stents in people with CHD.

Remember to include a justified conclusion.

[6 marks]

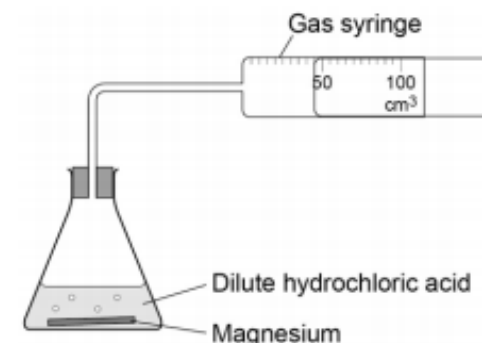
Evaluation – remember to
give balanced arguments
and a **conclusion**

0 3

A student investigated the rate of the reaction between magnesium and dilute hydrochloric acid.

The student used the apparatus shown in **Figure 4** to collect the gas produced.

Figure 4



0 3 . 1

Outline a plan to investigate how the rate of this reaction changed when the concentration of the hydrochloric acid was changed.

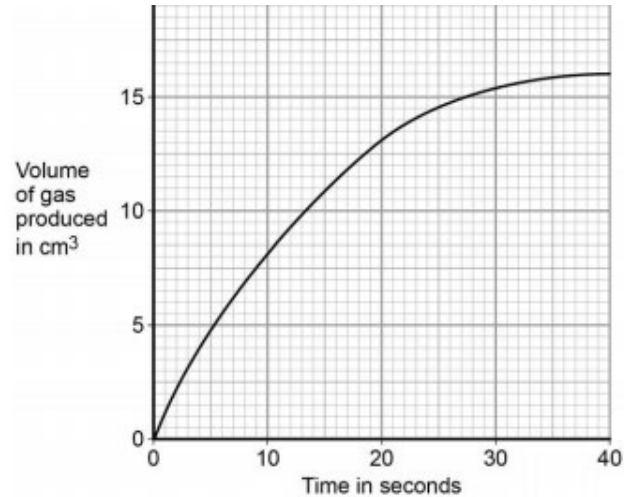
- Describe how you would do the investigation and the measurements you would make.
- Describe how you would make it a fair test.

You do **not** need to write about safety precautions.

[6 marks]

15% of GCSE marks in exams come
from questions relating to practicals.

DATA ANALYSIS AND EVALUATION



If data is given, use it!



Draw a tangent to the curve at 20 seconds.

Determine the rate of the reaction at 20 seconds by calculating the gradient of the tangent.

Give the unit.

[4 marks]

Rate of reaction = _____

Unit = _____

0 2 . 5

A driver wishes to buy a new car.

Table 1 gives some data about an electric car and one with a petrol engine.

Table 1

	Electric car	Petrol engine car
Cost (£)	27 000	15 000
Running cost per year (£)	250	2 000
Average lifetime (years)	12	12

Which car would be the most economic over its 12 year lifetime?

Use data from **Table 1** to support your answer.

You should include the difference in cost in your answer.

[4 marks]

THE EXAM PAPER

2

0 1

This question is about structure and bonding.

0 1 . 1

Figure 1 shows part of one layer of graphene.

Figure 1



*Do not write
outside the
box*

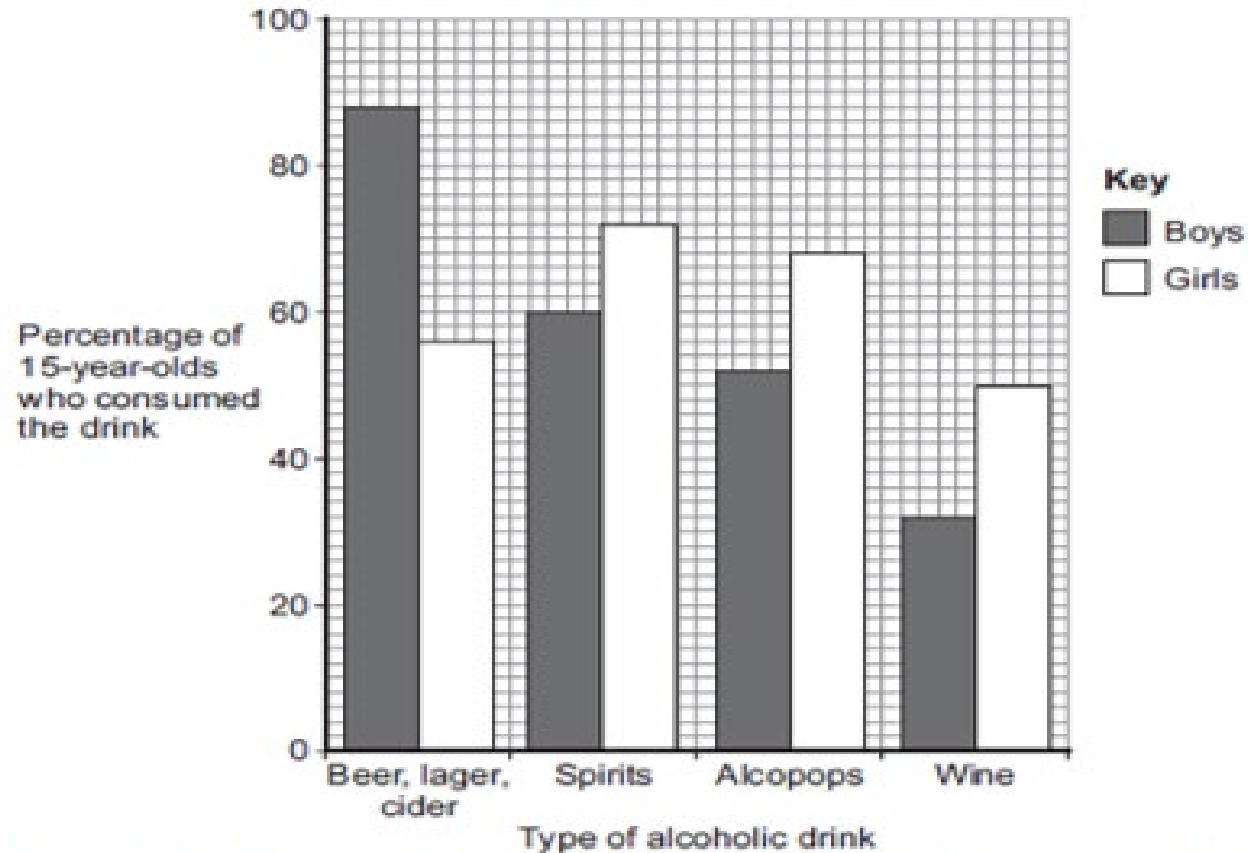
Do not write outside the box, exam papers are scanned and therefore any writing outside of the box may be missed and will not get marked

AQA - INSIGHT FROM THE LAST PREVIOUS EXAMS

Prepare

Figure 1 shows the results of a survey into the different types of alcoholic drinks consumed by one hundred 15-year-old boys and one hundred 15-year-old girls in the UK.

Figure 1



Describe the differences between the types of alcoholic drink consumed by boys and by girls.

Use only information from Figure 1.

[2 marks]

AQA - INSIGHT FROM THE LAST PREVIOUS EXAMS

Prepare for unfamiliar contexts

Read the question carefully too ensure you know what is being asked, understand the command words

Don't waste space repeating the question

Read through your work to check for errors

Be specific in your responses don't use 'it' or 'they'

Make sure you understand why each step in the practical is important

Maths - Show your working out in maths questions

- including the formula you are using if it isn't already given
- don't round answers until you reach the final answer

POSSIBLE TOPICS — COMBINED SCIENCE HIGHER

Biology 1

Cell specialization
Enzymes (PRAC)
Blood
Cancer
Drug development
Respiration

Biology 2

Contraception/infertility
Genetic inheritance
Genetic disorders
Evidence of evolution
Food chains
Biodiversity

Chemistry 1

Development of atom
Metallic bonding
States of matter
Polymers
Graphite/fullerenes
Metals reduction

Chemistry 2

Cracking
Waste water management
Reducing resources
Rate of reaction - temp

Physics 1

Energy changes - Kinetic,
gravitational potential,
elastic potential energy
Efficiency
National grid
Electricity generation

Physics 2

Work done
Energy transfers
Infra red radiation (PRAC)

POSSIBLE TOPICS — COMBINED SCIENCE FOUNDATION

Biology 1

Active transport
Enzymes (PRAC)
Blood
Cancer
Drug development
Respiration

Biology 2

Control of blood glucose
Contraception/infertility
Genetic disorders
Evolution and fossils
Genetic engineering
Classification
Maintaining biodiversity

Chemistry 1

Development of atom
Separating mixtures
Structure of diamond

Chemistry 2

Waste water management
Reducing resources
Rate of reaction - temp

Physics 1

Electricity generation
Radioactive contamination

Physics 2

Energy transfers
Infra red radiation (PRAC)

WHAT YOU CAN DO AS A PARENT - GET INVOLVED!

Help them **plan** their revision – small chunks

Question them using the revision guides

Mark the papers for them, the answers are available on exam board websites

Provide a calm environment..... remove distractions

Controlled access to electronic devices

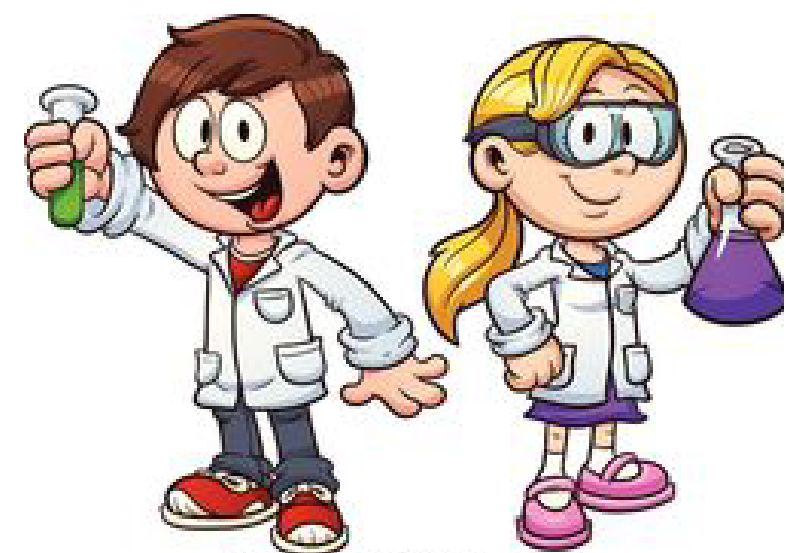
Food, water and exercise

Breakfast before exams!

Right equipment on the day..... Calculator!

Sleep!

ANY QUESTIONS



GOOD LUCK
in your
EXAMS!

You'll be AMAZING,
I asked around -
We all agreed!

